NOVEL SYNTHESIS OF 2,3-DIARYLBUTA-1,3-DIENES FROM 1,4-DIMETHOXYBUTYNE-2

Yoshio Ishino, Ikuzo Nishiguchi, Fujio Takihira, and Tsuncaki Hirashima Osaka Municipal Technical Research Institute, 2-1-1, Ogimachi, Kita-Ku, Osaka, Japan 530

Summary: It was found that 2,3-diarylbuta-1,3-dienes were readily obtained in good to excellent yields through the S_N^{-2} ' type substitution of 1,4-dimethoxybutyne-2 with aryl Grignard reagents in the presence of a copper(I) salt.

Recently 2-butyne-1,4-diol and its derivatives have received extensive attention because of their much availability¹⁾ and interesting reactivities²⁾, which have prompted us to explore useful application of these acetylenes to organic synthesis and effective industrial utilization.

In this study, we wish to describe a new synthetic method of 2,3-diarylbuta-1,3-dienes through the S_N^{-2} type substitution of 1,4-dimethoxybutyne-2 with aryl Grignard reagents in the presence of a copper(I) salt.

$$H_{3}COCH_{2}C^{\pm}CCH_{2}OCH_{3} + 2ArMgBr \xrightarrow{CuBr} CH_{2}=C \xrightarrow{C=CH_{2}} C=CH_{2}$$

ether Ar Ar
$$\underline{1} \qquad \underline{2} a-g \qquad \underline{3} a-g$$

It was well established that copper-catalyzed Grignard reagents³⁾ or cuprates⁴⁾ react with propargyl alcohol derivatives to give allenes. To our knowledge, however, none of one-pot synthesis of buta-1,3-dienes from acetylenic compounds has been reported.

A typical procedure is described as follow: To a solution of 0.03 mole of p-chlorophenylmagnesium bromide in 50 ml of anhydrous ether containing 0.45 g(0.003 mole) of cuprous bromide was added 1.14 g(0.01 mole) of 1,4-dimethoxybutyne- $2(\underline{1})^{2c}$ dissolved in 20 ml of anhydrous ether dropwise at 0 °C with vigorous stirring under nitrogen atmosphere, and was stirred at 0 °C for 2 hr and was gradually warmed to room temperature, and was allowed to stand at room temperature for 1 hr⁵⁾. After usual work-up of the mixture, 2,3-di(p-chlorophenyl)buta-1,3-dienes(<u>3</u>b) was isolated in a 98 % yield through column chromatograph; mp 64-65 °C, IR(KBr disk, cm⁻¹): $\mathcal{V}_{C=C}$ 1616, 1603, NMR(CCl₄, ppm): S = 5.40[doublet(J=2.0 Hz), 2H], 5.60[doublet(J=2.0 Hz), 2H], 7.30-7.42[multiplet, 8H], MASS: molecular peak m/e=274.

Under the similar conditions, the nucleophilic substitution of $\underline{1}$ with $\underline{2}$ in the presence of cuprous bromide⁶⁾, in principle, gave three types of products, namely 2,3-diarylbuta-1,3-dienes($\underline{3}$), 3-aryl-4-methoxybuta-1,2-dienes(4), and 1,3-diaryl-4-methoxybut-2-enes(5).

$$\frac{1}{2} + \frac{2}{ether} \xrightarrow{CuBr} CH_2 = C - C = CH_2 + CH_2 = C = C - CH_2OCH_3 + ArCH_2CH = CCH_2OCH_3$$

$$Ar Ar Ar Ar Ar Ar 5$$

All of the products were isolated and identified by spectroscopic methods(IR, NMR, MASS) and/or

Tab	le I The react	ion of 1 with 2	in the presence of c	uprous bromide*
Ar in ArMgBr			Isolated Yield (%)
	2	3	4	5
<u>2</u> a	C ₆ H ₅	66	+	19
<u>2</u> b	p-CIC ₆ H ₄	98	+	+
<u>2</u> c	m-C1C ₆ H ₄	100	0	0
<u>2</u> d	p-FC ₆ H ₄	81	8	7
<u>2</u> e	p-CH ₃ C ₆ H ₄	58	+	34
2f	$p-CH_3OC_6H_4$	70	+	7
<u>2g</u>	d ^{-c} 10 ^H 7	16	+	75
	* at 0 °C in et	ther + at	race amount	

elemental analyses. The results with each Grignard reagent are summarized in Table (⁷⁾

As shown in Table I, the introduction of an electron-withdrawing group into a phenyl ring of 2 brought about the exclusive formation of 3 in almost quantitative yields while an elecrtrondonating group on a phenyl ring of 2 led to the considerable increase in the yields of 5. Preferen tial formation of 5g to 3g in the reaction with 2g would be attributed to the steric hindrance of a bulky ϕ -naphthyl group on the attack of 2g to the inner carbon atom of 4g, the initially formed allenic ether.

2,3-Diarylbuta-1,3-dienes(3) are of much use in organic synthesis⁸⁾. For example, Diels-Alder reaction of 3c with dimethyl acetylenedicarboxylate yielded the adduct in a 77.2 % yield.

Finally. from the viewpoint of facile one-pot synthesis of a variety of buta-1,3-dienes from a common inexpensive acetylene derivatives, the present method may possess high potentiality in organic synthesis and industrial chemistry.

References and Notes

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- 5)3-Phenyl-4-methoxybuta-1,2-diene(4a) was obtained in a 88 % yield as a sole product when 1.50 equivalent moles of phenylmagnesium bromide (2a) was used in this nucleophilic substitution of 1. 6)Substitution of cuprous bromide with cuprous iodide or chloride as a catalyst in the reaction of 1 with 2a also led to the formation of 3a in 64 and 56 % yields, respectively.
- 7) The reaction of 1 with alkyl Grignard reagents in the presence of a copper(I) salt gave the corresponding 2,3-dialkylbutadienes and/or allenic ethers in low yield(20-48 %). Further study is on progress.
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